

## A New Nonaride Derivative, Dihydroepihevadride, as Characteristic Antifungal Agent against Filamentous Fungi, Isolated from Unidentified Fungus IFM 52672

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(Received for publication May 6, 2004)

In the screening of searching for new antifungal agents, a new nonaride compound, dihydroepihevadride (**1**), was isolated from unidentified fungus IFM 52672 as the most potent antifungal principle from this organism.

The structure of **1** was established on the basis of spectroscopic and chemical investigation, as well as detailed comparison of the spectroscopic and physico-chemical data of the oxidized derivative (**3**) from **1** with those of hevadride (**2**). Compound **1** showed strong antifungal activity against various filamentous fungi including human pathogens *Aspergillus fumigatus*, *Penicillium marneffeii* and *Trichophyton* spp. It also showed the growth inhibition activity against certain human pathogenic yeasts such as *Trichosporon* species, while it had weak or no antifungal activity against *Candida* spp. and *Cryptococcus neoformans*, and no antibacterial activity against *Bacillus subtilis* nor against *Escherichia coli*. The antifungal potencies of compounds **2** and **3** were found to be weaker than that of **1**.

The incidence of life-threatening fungal infections has steadily increased in immunocompromised hosts such as HIV infected persons and cancer and transplant patients.<sup>1)</sup> Invasive pulmonary aspergillosis and *Pneumocystis carinii* pneumonia are a leading cause of deaths in bone marrow transplant recipients and in HIV-infected patients, respectively. Moreover, resistance to the azoles, which are the most widely used antifungals today, is attracting much attention. Therefore, there is a continuing need for new antifungal agents to overcome these fungal diseases. In the screening for new antifungal substances from fungal sources against pathogenic filamentous fungi, *Aspergillus fumigatus* FRESENIUS IFM 41362 and *Aspergillus niger* VAN TIEGHEM H7160B, and/or pathogenic yeasts, *Candida albicans* (ROBIN) BERKHOUT IFM 40009 and *Cryptococcus neoformans* (SANFELICE) VUILLEMIN ATCC 90112, the chloroform-methanol (1:1) extracts of a freshly isolated unidentified fungus, IFM 52672, cultivated on rice for

21 days at 25°C, showed antifungal activity against *A. fumigatus* and *A. niger*.

### Results and Discussion

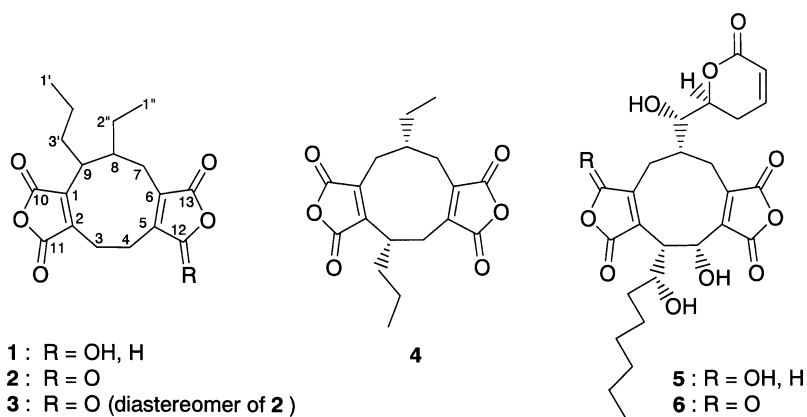
The molecular formula of compound (**1**) was determined as C<sub>18</sub>H<sub>22</sub>O<sub>6</sub> by high resolution electron-impact ionization mass spectrometry (HREI-MS). The physico-chemical properties and the <sup>1</sup>H- and <sup>13</sup>C-NMR data of **1** are shown in Tables 1 and 2, respectively. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **1** was similar to those of hevadride (**2**)<sup>2,3)</sup>, which was recently isolated from *Bipolaris heveae* CBS 241.93 (originally isolated from *Helminthosporium heveae* Petch<sup>2)</sup>), except for the extremely upfield shift of the carbon at C-12 ( $\delta$  98.7) in **1** instead of one of four carbonyl carbons in **2** and the new appearance of the <sup>1</sup>H-NMR signal at  $\delta$  6.04 (1H, s), corresponding to a proton attached to a

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Table 1. Physico-chemical properties of compounds 1~3.

	Dihydroepihevadride (1)	Heveadride (2)	Epihevadride (3)
Appearance	colorless crystalline powder (from diethyl ether)	colorless crystalline powder (from diethyl ether)	colorless crystalline powder (from diethyl ether)
Melting point (°C)	157-158	159.5-162	146-149
Molecular formula	C <sub>18</sub> H <sub>22</sub> O <sub>6</sub>	C <sub>18</sub> H <sub>20</sub> O <sub>6</sub>	C <sub>18</sub> H <sub>20</sub> O <sub>6</sub>
HREI-MS(M <sup>+</sup> )	334.1428 (Calcd. 334.1416)	332.1273 (Calcd. 332.1260)	332.1270 (Calcd. 332.1260)
CI-MS (isobutane)	335 (M+1)	333 (M+1)	333 (M+1)
IRν <sup>KBr</sup> cm <sup>-1</sup>	3450 (OH) 1840, 1770 (anhydride)	1834, 1761 (anhydride)	1830, 1763 (anhydride)
UV <sup>MeOH</sup> nm(log ε)	251 (3.56), 204 (4.18)	246 (3.30), 204 (3.67)	246 (3.94), 204 (4.29)
CD Δε(nm)	-5.76 (247)	+1.05 (220)	-13.16 (250)
[α] <sub>D</sub>	-91.3° (c=1.04, CHCl <sub>3</sub> )	+59.3° (c=1.12, CHCl <sub>3</sub> )	-170.9° (c=1.11, CHCl <sub>3</sub> )
Solubility (Soluble) (Insoluble)	CH <sub>2</sub> Cl <sub>2</sub> , EtOAc, Me <sub>2</sub> CO, EtOH, MeOH n-Hexane, H <sub>2</sub> O	CH <sub>2</sub> Cl <sub>2</sub> , EtOAc, Me <sub>2</sub> CO, EtOH, MeOH n-Hexane, H <sub>2</sub> O	CH <sub>2</sub> Cl <sub>2</sub> , EtOAc, Me <sub>2</sub> CO, EtOH, MeOH n-Hexane, H <sub>2</sub> O

Fig. 1. Structures of nonadrides and its derivatives.



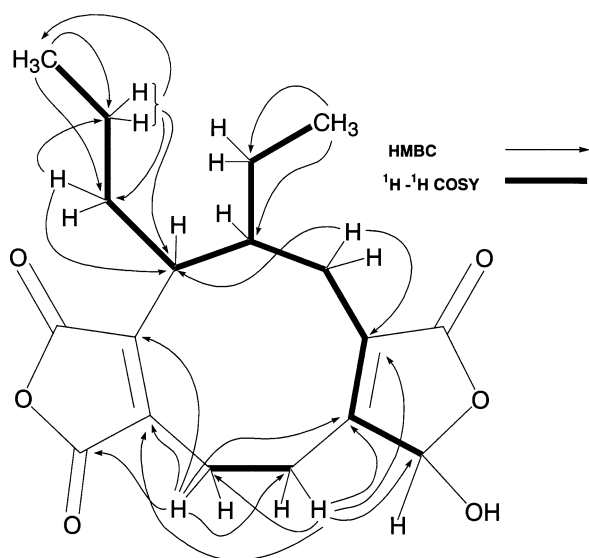
hemiacetal or acetal carbon (Table 2). The infrared (IR) absorption at 1840 cm<sup>-1</sup> (weak) and 1770 cm<sup>-1</sup> (strong) suggested the presence of a five-membered ring anhydride and a five-membered ester. These facts indicated that **1** could be the dihydro compound of **2**, *i.e.*, one of four carbonyls of two anhydride residues in **2** was reduced to a hemiacetal. From the detailed comparison of the <sup>1</sup>H- and <sup>13</sup>C-NMR signals of **1** with those of **2** and, from the analysis of the <sup>1</sup>H-<sup>1</sup>H COSY and HMBC spectra of **1** (Fig.

**2**), the planar structure of compound (**1**) was elucidated.

In order to determine the stereochemical relationship between **1** and **2**, **1** was oxidized with pyridium chlorochromate (PCC). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of compound (**3**) obtained was very similar to those of **2**, expect for the chemical shifts of the carbon signals at C-8 and C-9 (Table 2). The melting points, the optical rotation, and the CD value of **3** were different from those of **2** (Table 1). Therefore we concluded that **3** was a diastereomer of **2**,

Table 2.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectral data of dihydroepiheveadride (1), heveadride (2) and epiheveadride (3) in  $\text{CDCl}_3$ .

Carbon No.	Dihydroepiheveadride (1)		Heveadride (2)		Epiheveadride (3)	
	$^{13}\text{C}$	$^1\text{H}$	$^{13}\text{C}$	$^1\text{H}$	$^{13}\text{C}$	$^1\text{H}$
1	146.5		148.3		146.0	
2	146.5		145.1		146.2	
3	21.8	2.49 brt (12.5) 3.08 ddd (12.5, 10.0, 1.2)	21.2	2.34 brt (12.8) 3.08 brt (12.8)	21.5	2.20 m 3.16 m
4	24.3	2.36 m 2.73 m	22.5	2.20 m 3.23 dd (13.2, 8.6)	22.1	2.20 m 3.16 m
5	158.8		144.0		144.1	
6	130.6		144.5		145.0	
7	26.7	1.51m 2.64 brd (13.7)	26.1	2.12 brd (12.5) 2.88 brd (12.5)	28.0	1.76 brt (12.8, 12.5) 2.90 dd (12.8, 3.0)
8	46.5	2.04 m	40.7	2.22 m	47.9	2.10 m
9	40.1	2.64 m	46.2	2.24 m	40.8	2.44 m
10	164.5		163.5		164.1	
11	165.8		164.8		165.2	
12	98.7	6.04 m	165.7		165.2	
13	171.8		165.7		164.8	
1'	13.9	0.86 t (7.1)	14.0	0.86 t (7.0)	13.9	0.87 t (7.3)
2'	21.7	1.14 m 1.14 m	21.8	1.04 m 1.15 m	21.8	1.15 m 1.15 m
3'	31.5	1.51 m 2.14 m	30.8	1.62 m 1.94 m	31.6	1.56 m 2.15 m
1''	12.9	1.06 t (7.1)	12.8	1.18 brs	12.9	1.08 t (7.3)
2''	23.2	2.04 m 2.04 m	22.5	1.20 m 1.82 m	23.5	0.98 m 2.10 m

Fig. 2. HMBC and  $^1\text{H}$ - $^1\text{H}$  COSY correlations in dihydroepiheveadride (1).

designated as epiheveadride. The stereochemistry of **1** and **3** has not yet been determined.

More than ten nonadrides have been isolated from fungal sources, for example, byssochlamic acid (**4**) originally from *Byssoclomyces fulva* Olliver and Smith<sup>4)</sup> and *Byssoclomyces nivea*<sup>5)</sup> and heveadride (**2**) from *Helminthosporium heveae* Petch<sup>2)</sup>. However, there is only one example of a partially hydrogenated nonadride, which is rubratoxin A (**5**) originally isolated from *Penicillium rubrum* along with the typical nonadride, rubratoxin B (**6**)<sup>6)</sup>. Dihydroepiheveadride (**1**) is the second example of a partially hydrogenated nonadride to our knowledge.

#### Antimicrobial Property

Since none of the three compounds **1**~**3**, dissolve in water, the antimicrobial activity was determined by the paper disc method, as described in the previous paper<sup>7)</sup>. The results are summarized in Table 3. Compound **1** showed strong antifungal activity against various filamentous fungi including the human pathogens *A. fumigatus*, *P. marneffeii* and dermatophytes *T. rubrum* and *T. mentagrophytes* at a

Table 3. Antimicrobial activities of compounds 1~3.

Microorganisms	Diameter of Inhibition Zone (mm)					
	1		2		3	
	5 $\mu$ g	100 $\mu$ g	5 $\mu$ g	100 $\mu$ g	5 $\mu$ g	100 $\mu$ g
<Filamentous fungi>						
<i>Arthroderma benhamiae</i> IFM 41160	20		10		9	
<i>Aspergillus flavus</i> IFM 41935	26		—		—	
<i>Aspergillus fumigatus</i> IFM 41243	20		—		—	
<i>Aspergillus fumigatus</i> IFM 41362	20		—		—	
<i>Aspergillus fumigatus</i> IFM 47078	23		—		—	
<i>Aspergillus niger</i> IFM 41398	21		—		—	
<i>Cladophialophora carrionii</i> IFM 4808	—	26	—	—	—	—
<i>Emericella nidulans</i> IFM 46997	24		—		—	
<i>Epidermophyton floccosum</i> IFM 46637	13		9		8	
<i>Fonsecaea pedrosoi</i> IFM 4887		—		—		—
<i>Fusarium oxysporum</i> IFM 53787	10		—		—	
<i>Fusarium solani</i> IFM 52712	12		9		9	
<i>Microsporum canis</i> IFM 45108	25		10		9	
<i>Penicillium marneffeii</i> IFM 52703	21		—		—	
<i>Penicillium marneffeii</i> IFM 52697	11		—		—	
<i>Phialophora verrucosa</i> IFM 4928		—		—		—
<i>Scedosporium apiospermum</i> IFM 52028	17		11		11	
<i>Trichophyton mentagrophytes</i> IFM 40951	30		12		12	
<i>Trichophyton raubitschekii</i> IFM 45579	20		12		8	
<i>Trichophyton rubrum</i> IFM 45802	17		13		11	
<i>Trichophyton tonsurans</i> IFM 5275	28		12		9	
<i>Trichophyton verrucosum</i> IFM 46798	10		—		—	
<i>Trichophyton violaceum</i> IFM 46913	25		16		13	
<Yeasts>						
<i>Candida albicans</i> ATCC 90028	—	16	—	15	—	—
<i>Candida albicans</i> ATCC 90029	—	19	—	19	—	—
<i>Candida glabrata</i> IFM 40217	—	—	—	—	—	—
<i>Candida guilliermondii</i> IFM 46823	—	8	—	—	—	—
<i>Candida kefyr</i> IFM 46921	—	12	—	8	—	—
<i>Candida krusei</i> IFM 46834	—	15	—	—	—	—
<i>Candida parapsilosis</i> IFM 46863	—	—	—	—	—	—
<i>Candida tropicalis</i> IFM 46816	—	—	—	—	—	—
<i>Cryptococcus neoformans</i> ATCC 90112	—	14	—	11	—	9
<i>Saccharomyces cerevisiae</i> IFM 40210	—	20	—	—	—	—
<i>Pichia anomala</i> IFM 53788	—	—	—	—	—	—
<i>Trichosporon asahii</i> IFM 48429	15		10		—	
<i>Trichosporon asteroides</i> IFM 48608	10		—		—	
<Bacteria>						
<i>Bacillus subtilis</i> ATCC 6633	—		—		—	
<i>Escherichia coli</i> B	—		—		—	

Paper disc (i. d. 6 mm) was used.

The minus (—) means no inhibition.

concentration of 5  $\mu\text{g}/\text{disc}$ . In the activity test of **1** for yeasts at 5  $\mu\text{g}/\text{disc}$ , antifungal activity was observed on the two species, *Trichosporon asahii* and *Trichosporon asteroides*, but was not observed for such other human pathogens as *Candida albicans* and *Cryptococcus neoformans*. On the other hand, **1** inhibited the growth of several *Candida* species such as *C. albicans*, *C. krusei* and also *C. neoformans* at 100  $\mu\text{g}/\text{disc}$ . However, **1** did not show growth inhibition activity against of *Candida* species, *C. glabrata* and *C. tropicalis* nor dematiaceous fungi as *Fonsecaea pedrosoi* and *Phialophora verrucosa*.

The antifungal spectra of compounds **2** and **3** at 5  $\mu\text{g}/\text{disc}$  against the filamentous fungi were similar to that of **1**, but the potencies were lower.

For the yeasts, none of the three compounds show antifungal activity at 5  $\mu\text{g}/\text{disc}$ . Compounds **1** and **2** produced similar spectra at 100  $\mu\text{g}/\text{disc}$ , but the potency in **2** was relatively lower. On the other hand, no antifungal activity was observed in **3**, except for *C. neoformans*. Antibacterial activity of **1**~**3**, against *Bacillus subtilis* or *Escherichia coli* was not found.

These results have suggested that reduction of the carbonyl group at C-12 in nonadrides is more related to the intensity of the antifungal activity than stereochemical difference of side-chains at C-8 and C-9. The antifungal activity of **1** against filamentous fungi may contribute to discovery of a new molecular mechanism of action for antifungal agents.

## Experimental

Melting points were determined on a Yanagimoto micro-melting point apparatus and are uncorrected. EI- and CI-MS were taken with a JEOL JMS-MS600W spectrometer. UV and IR spectra were recorded on a Hitachi U-3210 spectrometer and a JASCO IR-810 spectrometer, respectively.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra were recorded on a JEOL Lambda-500 ( $^1\text{H}$ , 500.00 MHz;  $^{13}\text{C}$ , 125.25 MHz) spectrometer, using tetramethylsilane as an internal standard. CD curves were determined on a JASCO J-600 spectropolarimeter. Column chromatography was performed using Kieselgel 60 (Art. 7734, Merck) and Wakogel C-200 (Art. 237-00071, Wako). Low-pressure liquid chromatography (LPLC) was performed with a Chemco Low-Prep 81-M-2 pump and glass column (200 $\times$ 10 mm) packed with Silica gel CQ-3 (30~50  $\mu\text{m}$ , Wako). TLC were visualized by UV light at 254 nm and/or by spraying with 5%- $\text{H}_2\text{SO}_4$  and then heating.

### Antifungal Screening Assay of the Fungal Extracts.

The antifungal assay was performed by the paper disc method against *C. albicans* IFM 40009, and *C. neoformans* ATCC 90112, *A. fumigatus* IFM 41362, *A. niger* H7160B, as test organisms. About 220 source strains were cultured at 25°C for 21 days in a test-tube containing 10 g of moist rice. The cultivated rice was extracted with  $\text{CH}_2\text{Cl}_2$ -MeOH (1:1) and the organic layer was evaporated *in vacuo*. The extracts were spotted on the paper discs (8 mm diameter) at 2.5 mg/disc and placed on the assay plates. The fungi were cultivated in Potato Dextrose (PD) agar at 25°C. After 24~72 hours incubation, zones of inhibition (mm in diameter) were recorded.

### Isolation of Dihydroepihevadride (**1**) from Unidentified Fungus IFM 52672

The fungus IFM 52672 was cultivated on rice (500 g, using 5 Roux flasks) for 28 days at 25°C. The cultivated rice was extracted with  $\text{CHCl}_3$ -MeOH (1:1), and the evaporated extract (18 g) was suspended with water, and extracted with hexane (350 ml), benzene (350 ml),  $\text{CHCl}_3$  (350 ml), acetone (350 ml), and methanol (350 ml), sequentially. The hexane and benzene extracts showed antifungal activity against *A. fumigatus*. The combined residue of the hexane and benzene extracts was repeatedly chromatographed on silica gel (Wako, C-200) with benzene-chloroform, followed by recrystallization from diethyl ether to give compound **1** (315 mg), designated dihydroepihevadride. Physico-chemical properties and the  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR signal assignments of **1** are summarized in Tables 1 and 2, respectively.

### Oxidation of Dihydroepihevadride (**1**) by PCC (Pyridium Chlorochromate)

The  $\text{CH}_2\text{Cl}_2$  solution (3 ml) of dihydroepihevadride (**1**) (53 mg) was added to a  $\text{CH}_2\text{Cl}_2$  suspension (12 ml) of PCC (350 mg) and the mixture was refluxed for 2 hours with stirring. The reaction mixture was filtered, and the filtrate was evaporated *in vacuo*. The residue obtained was chromatographed on silica gel (Kieselgel 60, Merck) with  $\text{CH}_2\text{Cl}_2$ , followed by recrystallization from diethyl ether to give epihevadride (**3**) (41 mg). Physico-chemical properties of **3** are summarized in Table 1, and the  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR signal assignments are summarized in Table 2.

### Antibacterial and Antifungal Activities of Compounds **1**~**3**

Antibacterial and antifungal activities were *semi*-quantitatively determined using the agar diffusion method with paper disc (6 mm in diameter), loaded with 5 and/or

100  $\mu\text{g}$  of compound as described in the previous paper<sup>7)</sup>. The test organisms used and the results are summarized in Table 3.

#### Acknowledgements

We are grateful to Dr. H. KASAI and Miss N. KOBAYASHI of Hoshi University for NMR and mass measurements. This study was supported in part by Hoshi University Science/Technology Frontier Research Base from the Ministry of Education, Science, Sports and Culture, Japan and by a Cooperative Research Program of Research Center for Pathogenic Fungi and Microbial Toxicoses, Chiba University (04-23). This study was also performed as part of the program "Frontier Studies and International networking of Genetic Resources in Pathogenic Fungi and Actinomycetes (FN-GRPF)" through Special Coordination Funds for Promoting Science and Technology from the Ministry of Education, Culture, Sports, Science and Technology, the Japanese Government 2003.

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